

A Novel Route to Mixed Transition-Metal Complexes

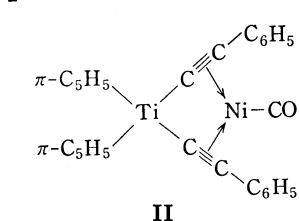
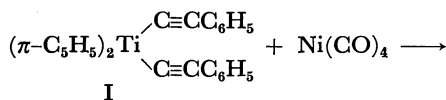
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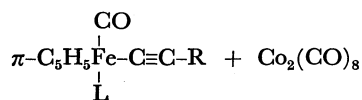
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Preparation of complexes containing different transition metals has recently attracted much attention.^{1,2)} We report herewith on the preparation of mixed transition metal complexes in which metal atoms are joined by the alkynyl group.

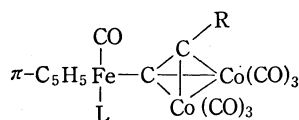
Bis- π -cyclopentadienyl-bisphenylethyntyttitanium (I)³⁾ reacted readily with nickel carbonyl in benzene at room temperature giving dark olive-green crystals (II) in good yield; mp 141–145°C (decomp.); NMR (CDCl₃, τ): 4.49 (s, C₅H₅) and 2.25–2.80 (m, C₆H₅); IR (KBr, ν , cm⁻¹): 1994 (vs, C \equiv O) and 1850 (C \equiv C coordinated). The data and the molecular weight (467.5 by vapor pressure osmometry and the molecular ion [M]⁺ at m/e 466 followed by [M-CO]⁺ at m/e 438 in the mass spectrum) suggest structure II.



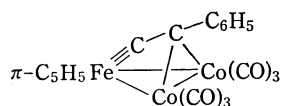
The propynyliron complex (IIIa)⁴⁾ reacted also readily with cobalt carbonyl giving black crystals



IIIa-c



-CO, L ↓



a (L: CO, R: CH₃)

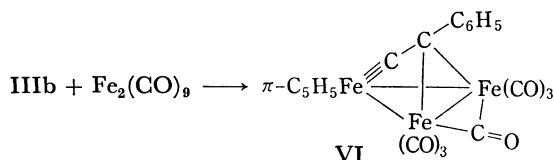
b (L: CO, R: C₆H₅)

c (L: PPhMe₂, R: C₆H₅)

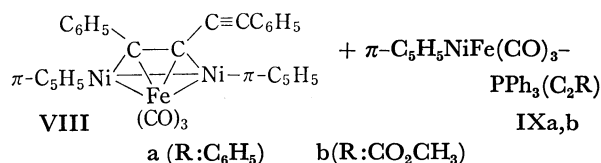
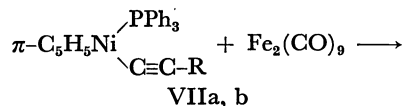
(IVa) in almost quantitative yield; mp 60–61°C; NMR: 5.01 (s, C₅H₅) and 7.32 (s, CH₃); IR (C₆H₁₄): 2065 (s), 2024 (vs), 1999 (s), and 1996 (sh) (terminal CO); Mass (m/e): 502 ([M]⁺) and ions [M- n CO]⁺ ($n=1\sim8$). The structure was assigned tentatively to IVa.

On the other hand, similar treatment of the phenylethyntyron complex (IIIb)⁵⁾ with the cobalt carbonyl afforded black crystals, C₁₉H₁₀O₆FeCo₂; mp 110–115°C (decomp.); Mass: 508 ([M]⁺) and ions [M- n CO]⁺ ($n=1\sim6$); IR: 2068 (s), 2026 (vs), 2000 (s), and 1986 (msh); NMR: 4.98 (s, C₅H₅) and 2.3–2.9 (m, C₆H₅). Structure V is postulated as a plausible one; unequivocal assignment being impossible with spectral data available at present. V might be formed via an intermediate IVb. Successful isolation of IVc, mp 91–94°C (decomp.), and its facile transformation into V support the above reaction sequence. Attempts to obtain a V analogue from IVa were not successful.

A similar triiron complex (VI), C₂₀H₁₀O₇Fe₃, mp 143–146°C, was obtained as brown crystals from the reaction of IIIb with iron nonacarbonyl in refluxing benzene; Mass: 530 ([M]⁺) and ions [M- n CO]⁺ ($n=1\sim7$); IR: 2053 (s), 2018 (vs), 2000 (s), 1996 (s), 1976 (s), and 1965 (s) (terminal CO) and 1870 (s, bridge CO).



Ethyntylnickel complexes (VII) reacted also with the iron carbonyl giving mixed metal complexes. Reaction of VIIa with carbonyl gave the known olive-green crystalline compound VIII⁶⁾ and a new complex (IXa), mp 195–197°C, as brown crystals. Similarly, the carbomethoxy derivative (IXb), mp 202–204°C, was obtained from VIIb. Analysis of the mass and IR spectra support formula IX.



a (R: C₆H₅) b (R: CO₂CH₃)

Thus, the transition metal acetylide complexes exhibit high reactivity toward metal carbonyls and provide a new type of mixed transition metal clusters.

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* Satisfactory microanalyses have been obtained for these compounds.

1) M. C. Baird, "Progress in Inorganic Chemistry," Vol. 9, ed. by F. A. Cotton, Interscience Publishers, N. Y. (1968), p. 1.

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